

**Atomic Energy Central School No.4 Rawatbhata
Multiple Choice Questions Test (April May 2019-20)**

MM: 100 Class XII (Physics, Chemistry, Biology) Time 90 Minutes

- 1 The rate of alpha particle falls on neutral sphere is 10^{12} per second. The time in which sphere gets charged by $2\mu\text{C}$ is sec.
A) 2.25 B) 3.15 C) 6.25 D) 1.66
- 2 Two point charges repel each other with a force of 100 N. One of the charges is increased by 10% and other is reduced by 10%. The new force of repulsion at the same distance would be..... N.
A) 121 B) 100 C) 99 D) 89
- 3 . Two small conducting sphere of equal radius have charges $+1\mu\text{C}$ and $-2\mu\text{C}$ respectively and placed at a distance d from each other experience force F_1 . If they are brought in contact and separated to the same distance, they experience force F_2 . The ratio of F_1 to F_2 is.....
A) $-8 : 1$ B) $1 : 2$ C) $1 : 8$ D) $-2 : 1$
- 4 Three charges, each of value Q , are placed at the vertex of an equilateral triangle. A fourth charge q is placed at the centre of the triangle. If the charges remain stationary then, $q =$
A) $Q/\sqrt{2}$ B) $Q/\sqrt{3}$ C) $-Q/\sqrt{2}$ D) $-Q/\sqrt{3}$
- 5 . Two equal negative charges $-q$ are fixed at points $(0, a)$ and $(0, -a)$. A positive charge Q is released from rest at the point $(2a, 0)$ on the X - axis. The charge Q will.....
A) move to the origin and remain at rest there B) execute simple harmonic motion about the origin C) move to infinity D) execute oscillations but not simple harmonic motion
- 6 Four charges, each equal to $-Q$, are placed at the corners of a square and a charge $+q$ is placed at its centre. If the system is in equilibrium, the value of q is
A) $Q/4(1+2\sqrt{2})$ B) $-Q/4(1+2\sqrt{2})$ C) $-Q/2(1+2\sqrt{2})$ D) $Q/2(1+2\sqrt{2})$
- 7 Two point positive charges q each are placed at $(-a, 0)$ and $(a, 0)$. A third positive charge q_0 is placed at $(0, y)$. For which value of y the force at q_0 is maximum
A) a B) $2a$ C) $a/\sqrt{2}$ D) $a/\sqrt{3}$
- 8 Two identical charged spheres suspended from a common point by two massless strings of length l are initially a distance d ($d \ll l$) apart because of their mutual repulsion. The charge begins to leak from both the spheres at a constant rate. As a result the spheres approach each other with a velocity u . Then function of distance x between them becomes
A) $u \propto x$ B) $u \propto x^{-1/2}$ C) $u \propto x^{-1}$ D) $u \propto x^{1/2}$
- 9 A charged particle of mass 1 kg and charge $2\mu\text{C}$ is thrown from a horizontal ground at an angle 45° with speed 20m/s. In space a horizontal electric field $E = 2 \times 10^7 \text{V/m}$ exist. The range on horizontal ground of the projectile thrown is
A) 100 meter B) 50 meter C) 200 meter D) 0 meter
- 10 A n electric dipole is placed at an angle of 60° with an electric field of intensity 10^5NC^{-1} . It experiences a torque equal to $8\sqrt{3}\text{Nm}$. If the dipole length is 2cm then the charge on the dipole is C.
A) -8×10^3 B) 8.54×10^{-4} C) 8×10^{-3} D) $.85 \times 10^{-6}$

- 11 A sphere of radius R has a uniform distribution of electric charge in its volume. At a distance x from its centre, (for $x < R$), the electric field is directly proportional to
- A) x B) x^{-1} C) x^{-2} D) x^2
- 12 Two Points P and Q are maintained at the Potentials of 10 v and -4 v, respectively. The work done in moving 100 electrons from P to Q is
- A) 2.24×10^{-16} J B) -9.60×10^{-17} J C) -2.24×10^{-16} J D) 9.60×10^{-17} J
- 13 The electric Potential V at any Point o (x, y, z all in metres) in space is given by $V = 4x^2$ volt. The electric field at the point (1m, 0.2m) in volt/metre is
- A) 8, along negative x - axis B) 8, along positives x - axis C) 16, along negative x - axis D) 16, along positives x - axis
- 14 Charges of $+3.33 \times 10^{-9}$ C are placed at each of the four corners of a square of side 8cm. The potential at the intersection of the diagonals is
- A) $150 \sqrt{2}$ Volt B) $900 \sqrt{2}$ Volt C) $1500 \sqrt{2}$ Volt D) 3600 Volt
- 15 . Two identical balls having like charges and placed at a certain distance apart repel each other with a certain force. They are brought in contact and then moved apart to a distance equal to half their initial separation. The force of repulsion between them increases 4.5 times in comparison with the initial value. The ratio of the initial charges of the balls is.....
- A) 4 : 1 B) 6 : 1 C) 3 : 1 D) 2 : 1
- 16 A point charge q is situated at a distance r from one end of a thin conducting rod of length L having a charge Q (uniformly distributed along its length). The magnitude of electric force between the two is.....
- A) $2kqQ/ r(r + L)$ B) $kqQ/ r(r + L)$ C) $kqQ/ r(r - L)$ D) $kQ/ r(r + L)$
- 17 Two point charges of $+16\mu\text{c}$ and $-9\mu\text{c}$ are placed 8 cm apart in air. Distance of a point from $-9\mu\text{c}$ charge at which the resultant electric field is zero.
- A) 24 cm B) 9 cm C) 16 cm D) 35 cm
- 18 An inclined plane making an angle of 30° with the horizontal is placed in an uniform electric field $E = 100 \text{ Vm}^{-1}$. A particle of mass 1 kg and charge 0.01 c is allowed to slide down from rest from a height of 1m. If the coefficient of friction is 0.2 the time taken by the particle to reach the bottom is sec.
- A) 2.337 B) 4.337 C) 5 D) 1.337
- 19 A small sphere whose mass is 0.1 gm carries a charge of 3×10^{-10} C of a silk fibre 5 cm long. The other end of the fibre is attached to a large vertical conducting, which has a surface charge of $25 \times 10^{-25} \text{ Cm}^{-2}$ on each side. When the system is freely hanging the angle fibre makes with vertical is.....
- A) 41.8° B) 45° C) 40.8° D) 45.8°
- 20 A Semicircular rod is charged uniformly with a total charge Q coulomb. The electric field intensity at the centre of curvature is.....
- A) $2KQ/ \pi r^2$ B) $3KQ/ \pi r^2$ C) $KQ/ \pi r^2$ D) $4KQ/ \pi r^2$
- 21 Two uniformly charged spherical conductors A and B having radius 1mm and 2mm are separated by a distance of 5 cm. If the spheres are connected by a conducting wire then in equilibrium condition, the ratio of the magnitude of the electric fields at the surfaces of spheres A and B is.....
- A) 4 : 1 B) 1 : 2 C) 2 : 1 D) 1 : 4
- 22 In Millikan's oil drop experiment an oil drop carrying a charge Q is held stationary by a p.d. 2400 v between the plates. To keep a drop of half the radius stationary the potential difference had to be made 600 v. What is the charge on the second drop?

- A) $3Q/2$ B) $Q/4$ C) Q D) $Q/2$
- 23 . Equal charges q are placed at the vertices A and B of an equilateral triangle ABC of side a . The magnitude of electric field at the point c is.....
- A) Kq/a^2 B) $3Kq/a^2$ C) $2Kq/a^2$ D) $q/\pi\epsilon_0 2a$
- 24 A Charge q is placed at the centre of the open end of cylindrical vessel. The flux of the electric field through the surface of the vessel is.....
- A) q/ϵ_0 B) $q/2\epsilon_0$ C) $2q/\epsilon_0$ D) Zero
- 25 An infinitely long thin straight wire has uniform linear charge density of $1/3 \text{ C/m}$. Then, the magnitude of the electric intensity at a point 18 cm away is..... NC^{-1}
- A) 0.66×10^{11} B) 1.32×10^{11} C) 0.33×10^{11} D) 3×10^{11}
- 26 . A long string with a charge of λ per unit length passes through an imaginary cube of edge l . The maximum possible flux of the electric field through the cube will be.....
- A) $\sqrt{3}\lambda/\epsilon_0$ B) λ/ϵ_0 C) $\sqrt{2}\lambda/\epsilon_0$ D) $6\lambda^2/\epsilon_0$
- 27 Three charges $2q, -q, -q$ are located at the vertices of an equilateral triangle. At the centre of the triangle.
- A) The Field is Zero but Potential is non - zero B) The Field is non - Zero but Potential is zero C) Both field and Potential are Zero D) Both field and Potential are non- Zero
- 28 In the electric field of a point charge q , a certain charge is carried from point A to B, C, D and E. Then the work done....
- A) Is least along the Path AB B) Is least along the Path AD C) Is Zero along all the Path AB, AC, and D) Is least along AE
- 29 . Three concentric spherical shells have radii a, b and c ($a < b < c$) and have surface charge densities $\sigma, -\sigma$ and σ respectively. If V_A, V_B and V_C denote the Potentials of the three shells, then for $c = a + b$, we have
- A) $V_C = V_B = V_A$ B) $V_C = V_B \neq V_A$ C) $V_C \neq V_B \neq V_A$ D) $V_C = V_A \neq V_B$
- 30 Two charged spheres of radii R_1 and R_2 having equal surface charge density. The ratio of their potential is ...
- A) R_2/R_1 B) $(R_2/R_1)^2$ C) $(R_1/R_2)^2$ D) R_1/R_2
- 31 If a charged spherical conductor of radius 10cm has potential v at a point distant 5 cm from its centre, then the potential at a point distant 15cm from the centre will be
- A) $1V/3$ B) $3V/2$ C) $3V$ D) $22V/3$
- 32 Electric potential at any point is $V = -5x + 3y + \sqrt{15}z$, then the magnitude of the electric field is N/C .
- A) $3\sqrt{2}$ B) $4\sqrt{2}$ C) 0 D) $5\sqrt{2}$
- 33 . A simple pendulum of period T has a metal bob which is negatively charged. If it is allowed to oscillate above a positively charged metal plate, its period will.....
- A) Remains equal to T B) Less than T C) Infinite D) Greater than T
- 34 Which of the following units is useful in relating concentration of solution with its vapour pressure?
- A) Mole fraction B) ppm C) Mass percentage D) Molality
- 35 Maximum amount of a solid solute that can be dissolved in a specified amount of a given liquid solvent does not depend upon:
- A) Temperature B) Nature of solute C) Pressure D) Nature of solvent
- 36 Which of the following aqueous solutions should have the highest boiling point?

- A) 1M NaOH B) 1M Na₂SO₄ C) 1M NH₄NO₃ D) 1M KNO₃
- 37 The value of Van't Hoff factors for KCl, NaCl and K₂SO₄ respectively are:
 A) 2,2 & 2 B) 2,2 & 3 C) 1,1 & 2 D) 1,1 & 1
- 38 Value of Henry's constant K_H is:
 A) Increase with increase in temperature B) Decrease with increase in temperature C) Remains constant D) First increase then decrease
- 39 The boiling point of a solvent containing a non-volatile solute:
 A) Is depressed B) Is elevated C) Does not change D) none
- 40 The molality of pure water is:
 A) 55.5 B) 20 C) 18 D) 10
- 41 Which of the following concentrations is not affected by temperature:
 A) Normality B) Molality C) Molarity D) Formality
- 42 The number of moles of NaCl in 3 litres of 3M solution is:
 A) 1 B) 3 C) 9 D) 27
- 43 The amount of solute required to prepare 10 litres of decimolar solution is:
 A) 0.01mole B) 0.2mole C) 0.05mole D) 1.0mole
- 44 One kilogram of water contains 4.0g of NaOH. The concentration of the solution is best expressed as:
 A) 0.1molal B) 0.1molar C) decinormal D) About 0.1mole
- 45 Isotonic solutions are the solution having the same:
 A) Surface tension B) Vapour pressure C) Osmotic pressure D) Viscosity
- 46 Which of the following is not a colligative property?
 A) Depression in FP B) Elevation in BP C) Osmotic pressure D) Lowering of vapour pressure
- 47 An aqueous solution containing 6.0 g of urea in 500 ml of solution has a density equal to 1.05 g/cm³. If the molar mass of urea is 60, then molality of solution is:
 A) 0.20 B) 0.19 C) 0.10 D) 1.20
- 48 Which pair will not form an ideal solution?
 A) C₂H₅Br & C₂H₅I B) C₆H₅Br & C₆H₅I C) C₆H₆ & C₆H₅CH₃ D) C₂H₅I & C₂H₅OH
- 49 The Van't Hoff factor for 0.1 M Ba(NO₃)₂ solution is 2.74. The degree of dissociation is:
 A) 91.3% B) 87% C) 100% D) 74%
- 50 Camphor is often used in molecular mass determination because:
 A) High cryoscopic constant B) It is volatile C) It is solvent for organic substances D) It is readily available

- 51 The mole fraction of methanol in its 4.5 molal aqueous solution is:
 A) 0.250 B) 0.125 C) 0.100 D) 0.075
- 52 The difference between the electrode potentials of two electrodes when no current is drawn through the cell is called:
 A) Cell potential B) Cell emf C) Potential difference D) Cell voltage
- 53 An electrochemical cell can behave like an electrolytic cell when:
 A) $E_{\text{cell}} = 0$ B) $E_{\text{cell}} > E_{\text{ext}}$ C) $E_{\text{ext}} > E_{\text{cell}}$ D) $E_{\text{cell}} = E_{\text{ext}}$
- 54 The quantity of charge required to obtain one mole of aluminium from Al_2O_3 is:
 A) 1F B) 6F C) 3F D) 2F
- 55 Molar conductivity of ionic solution depends on:
 A) Pressure B) Distance between electrodes C) Concentration of solution D) Surface area of electrodes
- 56 The SI unit of molar conductivity is:
 A) $\text{S m}^2 \text{mol}^{-1}$ B) $\text{S m}^{-1} \text{mol}^{-1}$ C) $\text{S m}^2 \text{mol}$ D) $\text{S m}^3 \text{mol}^{-1}$
- 57 If the conductivity and conductance of a solution is same then its cell constant is equal to:
 A) 1 B) 0 C) 10 D) 1000
- 58 E°_{cell} and ΔG° are related as:
 A) $\Delta G^{\circ} = nFE^{\circ}_{\text{cell}}$ B) $\Delta G = -nFE^{\circ}_{\text{cell}}$ C) $\Delta G^{\circ} = -nFE^{\circ}_{\text{cell}}$ D) $\Delta G^{\circ} = -nFE^{\circ}_{\text{cell}} = 0$
- 59 Rust is a mixture of :
 A) FeO & $\text{Fe}(\text{OH})_3$ B) FeO & $\text{Fe}(\text{OH})_2$ C) Fe_2O_3 & $\text{Fe}(\text{OH})_3$ D) Fe_3O_4 & $\text{Fe}(\text{OH})_3$
- 60 The emf of the cell $\text{Cu}_{(\text{s})} | \text{Cu}^{2+} (1\text{M}) || \text{Ag}^+ (1\text{M}) | \text{Ag}$ is 0.46 V. The standard REP of Ag^+/Ag is 0.80 V. The standard REP of Cu^{2+}/Cu is:
 A) -0.34 V B) 1.26 V C) -1.26 V D) 0.34 V
- 61 Consider the following E° values, $E^{\circ}(\text{Fe}^{3+}|\text{Fe}^{2+}) = +0.77 \text{ V}$, $E^{\circ}(\text{Sn}^{2+}|\text{Sn}) = -0.14 \text{ V}$. Under standard conditions, the potential for the reaction:
 $\text{Sn}_{(\text{s})} + 2\text{Fe}^{3+}_{(\text{aq})} \rightarrow 2\text{Fe}^{2+}_{(\text{aq})} + \text{Sn}^{2+}_{(\text{aq})}$ is:
 A) 0.91 V B) 1.04 V C) 1.68 V D) 0.63 V
- 62 The limiting molar conductivities λ° for NaCl , KBr and KCl are 126, 152 and 150 $\text{S cm}^2 \text{mol}^{-1}$ respectively. The λ° for NaBr is:
 A) 278 $\text{S cm}^2 \text{mol}^{-1}$ B) 976 $\text{S cm}^2 \text{mol}^{-1}$ C) 128 $\text{S cm}^2 \text{mol}^{-1}$ D) 302 $\text{S cm}^2 \text{mol}^{-1}$
- 63 For spontaneity of a cell, which is correct?
 A) $\Delta G = 0$, $E^{\circ} = 0$ B) $\Delta G = -ve$, $E^{\circ} = 0$ C) $\Delta G = +ve$, $E^{\circ} = 0$ D) $\Delta G = -ve$, $E^{\circ} = +ve$
- 64 A hydrogen gas electrode is made by dipping platinum wire in a solution of HCl of $\text{pH} = 10$ and by passing hydrogen gas around the platinum wire at one atm pressure. The oxidation potential of electrode would be?
 A) 0.118 V B) 1.18V C) 0.059 V D) 0.59 V

- 65 When 0.1 mol MnO_4^{2-} is oxidized, the quantity of electricity required to completely oxidize MnO_4^{-1} is:
 A) 96500 C B) 2x96500 C C) 9650 C D) 96050 C
- 66 When the same quantity of electricity is passed for half an hour, the amount of Cu and Cr deposited are 0.375 and 0.30 respectively. Ratio of electrochemical equivalent of Cu and Cr is:.
 A) 0.8 B) 1.25 C) 2.5 D) 1.62
- 67 The residual persistent nucellus is known as:
 A) Perisperm B) Pericarp C) Integuments D) None of these
- 68 Which one from the following statement is incorrect:
 A) In birds and mammal internal fertilization takes place
 B) Colostrums contains antibodies and nutrients
 C) Polyspermy is prevented by the chemical changes in the egg surface
 D) In the human female implantation occurs almost seven days after fertilization
- 69 Identify the odd one from the following:
 A) Labia minora B) Fimbriae C) Infundibulum D) Isthmus
- 70 Division of human egg is:
 A) Holoblastic, equal B) Meroblastic C) Holoblastic, unequal D) None of these
- 71 The numbers of chromosomes in mature gamete gets halved during:
 A) Meiosis II B) Formation of first polar body
 C) Formation of second polar body D) Division of secondary oocyte and spermatocyte
- 72 Plants with ovaries having one or a few ovules, are generally pollinated by:
 A) Bees B) Butterflies C) Birds D) Wind
- 73 Ovulation in the human female normally takes place during the menstrual cycle:
 A) At the beginning of proliferative phase B) At the end of proliferative phase C) At the mid of secretory phase D) Just before the end of secretory
- 74 Acrosome is a type of:
 A) Lysosome B) Flagellum C) Ribosome D) Basal body
- 75 Which part of ovary in mammals acts as an endocrine gland after ovulation:
 A) Vitelline membrane B) Graffian follicle C) Stroma D) Germinal epithelium
- 76 Withdrawal of which of the following hormones is immediate cause of menstruation?
 A) Estrogen B) FSH C) FSH-RH D) Progesterone
- 77 Layers of an ovum from outside to inside are:

- A) Corona radiata, zona pellucida and vitelline membrane
 B) Zona pellucida, corona radiata and vitelline membrane
 C) Vitelline membrane, zona pellucida and corona radiata
 D) Zona pellucida, vitelline membrane and corona radiata
- 78 Which one of the following is the most likely reason of not occurring regular menstruation cycle in females?
- A) Fertilization of ovum
 B) Maintenance of hypertrophical endometrial lining
 C) Retention of corpus luteum
 D) Maintenance of high concentration of sex hormones in blood
- 79 The first movement of the foetus and appearance of hair on its head are usually observed during which month of pregnancy?
- A) Four month B) Fifth month C) Sixth month D) Third month
- 80 The coconut water from tender coconut represents:
- A) Free nuclear proembryo. B) Free nuclear endosperm. C) Endocarp. D) Fleshy mesocarp.
- 81 Change in GnRH pulse frequency in females is controlled by circulating levels of:
- A) Progesterone only B) Progesterone and inhibin C) Estrogen and progesterone D) Estrogen and inhibin
- 82 Seed formation without fertilization in flowering plants involves the process of:
- A) Somatic hybridization B) Apomixes C) Sporulation D) Budding
- 83 Which of the following statements is not correct?
- A) LH and FSH decrease gradually during the follicular phase.
 B) LH triggers secretion of androgens from the Leyding cells
 C) FSH stimulates the sertolli cells which help in spermiogenesis
 D) LH triggers ovulation in overy
- 84 Which one of the following pairs is wrongly matched, while the remaining three are correct?
- A) *Bryophyllum* –Leaf buds B) *Penicilium* --- Conidia C) *Agave* --- Bulbils D) *Water hyacinth*--- Runner
- 85 Vegetative propagation in *Pistia* occurs by:
- A) Stolon B) Offset C) Runner D) Sucker
- 86 Both autogamy and geitonohgamy are prevented in:
- A) Papaya B) Cucumber C) Caster D) Maize
- 87 Vegetative propagation in mint occurs by:
- A) Runner B) Offset C) Rhizome D) Sucker

- 88 *Hydra* reproduced asexually through:
- A) Fragmentation B) Budding C) totipotancy D) Pleutipotency
- 89 During oogenesis, each diploid oocyte produces:
- A) Four functional ova B) Two functional eggs and two polar bodies
C) Four functional polar bodies D) One functional egg and three polar bodies
- 90 The fleshy edible part of an apple is:
- A) Thalamus B) Nucellus C) Ovary D) Endosperm
- 91 What is common between vegetative reproduction and apomixes?
- A) Both occur round the year B) Both produce progeny identical to parent
C) Both are applicable to only dicot plants D) Both bypass the flowering plant
- 92 The "eyes" of the potato tubers are:
- A) Root buds B) Flower buds C) Shoot buds D) Axillary buds
- 93 In angiosperms, functional megaspore develop into:
- A) Endosperm B) Pollen sac C) Embryo sac D) Ovule
- 94 Which one of the following is surrounded by a callose wall?
- A) Male gamete B) Egg C) Pollen grain D) Microspore mother cell
- 95 What is the function of germ pore:
- A) Absorption of water for seed germination B) Initiation of pollen tube C) Release of male gametes D) Meiosis II
- 96 The fertilization in which male gamete are carried though pollen tube is known as:
- A) Syngamy B) Porogamy C) Siphonogamy D) Chalazogmy
- 97 In which one of the following pollination is autogamous?
- A) Geitonogamy B) Xenogamy C) Chasmogamy D) Cleistogamy
- 98 Morula is a developmental stage:
- A) Between implantation and parturition B) Between the zygote and blastocyte C) Between the blastocyte and gastrula D) After implantation
- 99 Which of the following is hormonal releasing intra uterine devices (IUD)
- A) multoload-375 B) LNG -20 c) cervical cap D) vault
- 100 Shortest life span of
- A) Mayfly B) human C) Housefly D) fruitfly

Answer Key XII PCB

1	2	3	4	5	6	7	8	9	10
C	C	A	B	D	B	C	B	C	C
11	12	13	14	15	16	17	18	19	20
A	A	A	C	D	B	A	D	C	A
21	22	23	24	25	26	27	28	29	30
C	D	C	D	C	A	B	C	D	D
31	32	33	34	35	36	37	38	39	40
D	C	B	A	C	B	B	A	B	A
41	42	43	44	45	46	47	48	49	50
B	C	D	A	C	D	B	D	B	A
51	52	53	54	55	56	57	58	59	60
D	B	C	C	C	A	A	C	C	D
61	62	63	64	65	66	67	68	69	70
A	C	D	D	C	B	A	c	A	A
71	72	73	74	75	76	77	78	79	80
B	D	b	A	B	D	A	A	b	B
81	82	83	84	85	86	87	88	89	90
C	B	B	D	B	A	D	B	D	A
91	92	93	94	95	96	97	98	99	100
B	C	C	d	B	C	D	B	b	A